

From the INTERNATIONAL PRELIMINARY EXAMINING AUTHORITY

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NOTIFICATION OF TRANSMITTAL OF THE INTERNATIONAL PRELIMINARY **EXAMINATION REPORT** 

(PCT Rule 71.1)

Date of mailing

(day/month/year)

23.02.2005

Applicant's or agent's file reference

2001B124A

IMPORTANT NOTIFICATION

International application No. PCT/US 03/18713

International filing date (day/month/year) 12.06.2003

Priority date (day/month/year)

12.12.2002

Applicant

EXXONMOBIL CHEMICAL PATENTS, INC. et al.

- The applicant is hereby notified that this International Preliminary Examining Authority transmits herewith the international preliminary examination report and its annexes, if any, established on the international application.
- 2. A copy of the report and its annexes, if any, is being transmitted to the International Bureau for communication to all the elected Offices.
- 3. Where required by any of the elected Offices, the International Bureau will prepare an English translation of the report (but not of any annexes) and will transmit such translation to those Offices.

#### 4. REMINDER

The applicant must enter the national phase before each elected Office by performing certain acts (filing translations and paying national fees) within 30 months from the priority date (or later in some Offices) (Article 39(1)) (see also the reminder sent by the International Bureau with Form PCT/IB/301).

Where a translation of the international application must be furnished to an elected Office, that translation must contain a translation of any annexes to the international preliminary examination report. It is the applicant's responsibility to prepare and furnish such translation directly to each elected Office concerned.

For further details on the applicable time limits and requirements of the elected Offices, see Volume II of the PCT Applicant's Guide.

The applicant's attention is drawn to Article 33(5), which provides that the criteria of novelty, inventive step and industrial applicability described in Article 33(2) to (4) merely serve the purposes of international preliminary examination and that "any Contracting State may apply additional or different criteria for the purposes of deciding whether, in that State, the claimed inventions is patentable or not" (see also Article 27(5)). Such additional criteria may relate, for example, to exemptions from patentability, requirements for enabling disclosure, clarity and support for the claims.

> ACKNOWLEDGED PATENT LEGAL ASSISTANT GROUP

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# **PCT**

## INTERNATIONAL PRELIMINARY EXAMINATION REPORT

(PCT Article 36 and Rule 70)

Applicant's or agent's file reference 2001B124A  International application No. PCT/US 03/18713			FOR FURTHER ACT	FOR FURTHER ACTION  See Notification of Transmittal of International Preliminary Examination Report (Form PCT/IPEA/416)			
			International filing date (da 12.06.2003	y/month/year)	Priority date (day/month/year) 12.12.2002		
C08F2	291,00	ent Classification (IPC) o	r both national classification and	IPC			
Applicar EXXO		L CHEMICAL PATE	NTS, INC. et al.	<u>-</u>			
1. T	<ol> <li>This international preliminary examination report has been prepared by this International Preliminary Examining Authority and is transmitted to the applicant according to Article 36.</li> </ol>						
2. T	2. This REPORT consists of a total of 4 sheets, including this cover sheet.						
×	This report is also accompanied by ANNEXES, i.e. sheets of the description, claims and/or drawings which have been amended and are the basis for this report and/or sheets containing rectifications made before this Authority (see Rule 70.16 and Section 607 of the Administrative Instructions under the PCT).						
ТІ	hese an	nexes consist of a tota	ll of 3 sheets.				
3. TI	his repo	rt contains indications	relating to the following item	s:			
		Basis of the opinion					
· I <del>I</del>		Priority					
 HI			of oninion with regard to nove	elty inventive	step and industrial applicability		
١٧		Lack of unity of inve	•	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	o otop and industrial approaching		
V		Reasoned statemen			relty, inventive step or industrial applicability;		
V	ı 🗆	Certain documents	cited				
V	II 🗆	Certain defects in th	e international application				
Vi	III 🗇	Certain observations	on the international applica	tion			
Date of s	submissio	on of the demand	D	ate of complet	ion of this report		
08.06.2004			2	3.02.2005			
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# INTERNATIONAL PRELIMINARY EXAMINATION REPORT

International application No.

PCT/US 03/18713

I.	<b>Basis</b>	of the	report
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**Description, Pages** 

1. With regard to the **elements** of the international application (Replacement sheets which have been furnished to the receiving Office in response to an invitation under Article 14 are referred to in this report as "originally filed" and are not annexed to this report since they do not contain amendments (Rules 70.16 and 70.17)):

	1, 3, 5-11, 13-31		as originally filed
	2, 4	l, 12	received on 27.01.2005 with letter of 27.01.2005
	Cla	ims, Numbers	
	1-3	6	as originally filed
	Dra	ıwings, Sheets	
	1/8-	·8/8	as originally filed
2.	Witi lang	h regard to the <b>langu</b> guage in which the int	age, all the elements marked above were available or furnished to this Authority in the ternational application was filed, unless otherwise indicated under this item.
	The	ese elements were av	ailable or furnished to this Authority in the following language: , which is:
		the language of a tra	anslation furnished for the purposes of the international search (under Rule 23.1(b)).
		the language of publ	lication of the international application (under Rule 48.3(b)).
		the language of a tra Rule 55.2 and/or 55.3	anslation furnished for the purposes of international preliminary examination (under 3).
3.			otide and/or amino acid sequence disclosed in the international application, the examination was carried out on the basis of the sequence listing:
		contained in the inter	rnational application in written form.
		filed together with the	e international application in computer readable form.
		furnished subsequen	ntly to this Authority in written form.
		furnished subsequen	ntly to this Authority in computer readable form.
		The statement that the in the international approximation of the international approximation of the statement of the statemen	he subsequently furnished written sequence listing does not go beyond the disclosure pplication as filed has been furnished.
		The statement that the listing has been furni	he information recorded in computer readable form is identical to the written sequence ished.
ŧ.	The	amendments have re	esulted in the cancellation of:
		the description,	pages:
		the claims,	Nos.:
		the drawings,	sheets:

#### INTERNATIONAL PRELIMINARY **EXAMINATION REPORT**

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5.	This report has been established as if (some of) the amendments had not been made, since they have
	been considered to go beyond the disclosure as filed (Rule 70.2(c)).

(Any replacement sheet containing such amendments must be referred to under item 1 and annexed to this report.)

6. Additional observations, if necessary:

V. Reasoned statement under Article 35(2) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement

1. Statement

Novelty (N)

No:

Yes: Claims Claims 12-36

Inventive step (IS)

Yes: Claims

1-11 12-36

No: Claims

Industrial applicability (IA)

Yes: Claims

1-36

Claims No:

2. Citations and explanations

see separate sheet

- D1: EP-A-1 050 548 (CHEMPLAST MARKETING SERVICES E) 8 November 2000 (2000-11-08)
- D2: EP-A-0 507 075 (BASF AG) 7 October 1992 (1992-10-07)
- D3: EP-A-0 113 259 (CEAT STE FSE) 11 July 1984 (1984-07-11)
- D4: US-A-5 169 900 (GUDELIS PAUL V) 8 December 1992 (1992-12-08)
- D5: US-A-5 741 858 (CREE STEPHEN H ET AL) 21 April 1998 (1998-04-21)
- D2 is considered the closest prior art document. 1. According to the Applicants, D2 does not teach vulcanization or final cross linking with water of the silane grafted resilient polymer as a separate step. In support of this they refer to claims 12 and 20 to 23 as originally filed. These claims are not suggested from the contents of D2 (Art. 33(3) PCT).
- 2. It is clear from the above that the presence of water is an essential feature which must be incorporated into the main claim 1 in order to make it clear what is actually claimed.
  - The objections against claims 1 to 11 as given in the official opinion must be maintained as long as the novelty feature "water" is not present in these claims unless there is a further novelty feature in these claims 1 to 11 which is hitherto unknown (Art. 33(2) PCT).
- 3. D1 and D2 are acknowledged at new page 2. The other documents cited are only background art.

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#### REPLACEMENT SHEETS

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polypropylene matrix. In a first step, the ethylene-octene polymers are coated and peroxide generation upon melting causes grafting (See *Polymer Engineering and Science* at page 1092). The polypropylene appropriately coated is added and blended in a second step. In a third step, water is injected to effect cross-linking. DE4402943 discloses a similar process.

[0007] PCT publication WO 98/23687 discloses thermoplastic polymer blend compositions that include a thermoplastic matrix resin phase that is substantially free of cross-linking and a dispersed silane-grafted elastomer phase.

[0008] It is among the objects of the invention to provide a simplified and/or flexible process by integrating blending and grafting and/or blending and curing. EP-A-1 050 548 and EP-A-0 507 075 represent prior art grafted thermoplastic compositions.

### 3. SUMMARY OF THE INVENTION

15 [0009] In one embodiment, the present invention provides a process for making a thermoplastic vulcanizate ("TPV") in a reactor. The process includes forming a mixture in which a silane grafted resilient polymer component is dispersed in a thermoplastic matrix component and adding a solid water-generating agent to crosslink the silane grafted elastomer component. The mixture is formed by mixing in the reactor: a) from 40 to 75 parts by weight of the matrix component, per 100 parts by weight of the matrix component and resilient polymer component combined, b) from 25 to 60 parts by weight of the resilient polymer component, per 100 parts by weight of the matrix component and resilient polymer component combined, and c) a silane grafting agent.

25 [0010] In another embodiment, the process includes a) blending a thermoplastic polymer component for forming a continuous matrix phase, a resilient polymer component, and a silane grafting agent for forming a phase dispersed in the matrix, and additives so as to promote silane grafting; and b) adding a solid water generating agent, which releases water, while the blend formed in step a) is submitted to shear so as to crosslink the silane grafted polymer.

[0011] In a particular aspect of any of the embodiments described herein, the process has one or more of the following characteristics, in any combination:

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### REPLACEMENT SHEET

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- q) the reactor is a continuous-type compounding apparatus;
- r) the reactor is connected to a die suitable for extruding the product in the reactor into a shaped, fabricated product without an intervening pelletization step;
- 5 s) the matrix component has a crystallinity as determined by DSC of at least 40%;
  - t) the resilient polymer component has a crystallinity as determined by DSC of no more than 40%;
  - u) the crystallinity of the matrix component and the resilient polymer component differ by at least 10%, or at least 20%; and
  - v) the matrix component and the resilient polymer component are blended and simultaneously combined with the silane grafting agent.

# 4. BRIEF DESCRIPTION OF THE DRAWINGS

- 15 [0012] Figure 1 is a graph of the thermogravimetric analysis of weight loss versus temperature for magnesium sulfate heptahydrate (Epsom salt).
  - [0013] Figure 2 is a graph of the thermogravimetric analysis of weight loss versus temperature for sodium sulfate decahydrate (Glauber's salt).
  - [0014] Figure 3 is a graph of the thermogravimetric analysis of weight loss versus temperature for talc.
  - [0015] Figure 4 is a graph of the thermogravimetric analysis of weight loss versus temperature for hydrated clay (hydrous aluminum silicate).
  - [0016] Figure 5 is the thermogravimetric analysis of weight loss versus temperature for a zinc oxide/stearic acid combination.
- 25 [0017] Figure 6 is the thermogravimetric analysis of weight loss versus temperature for a zinc oxide/isononanoic acid combination.
  - [0018] Figure 7 is the thermogravimentic analysis of weight loss versus temperature for a zinc oxide/isooctanoic acid combination.
  - [0019] Figure 8 is a low voltage SEM micrograph of calendared sheeting.

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#### REPLACEMENT SHEET

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[0042] When the silane grafting reaction is complete, a water-generating agent releases water upon heating, and preferably at the melting temperature range of the polymers, inside the compounding equipment, which enables the crosslinking to occur. The water-generating agent can be added to the reactor upon completion of the silane grafting reaction. Examples of water-generating agents include Epsom salt, Glauber's salt, clay, water, talc, metal oxide/carboxylic acid combinations, and combinations thereof. Examples of metal oxide/carboxylic acid combinations are zinc oxide/stearic acid, zinc oxide/isononaioc acid, and zinc oxide isooctanoic acid.

[0043] Figures 1 and 2 illustrate the thermogravimetric analysis of weight loss versus temperature for magnesium sulfate heptahydrate (Epsom salt), and sodium sulfate decahydrate (Glauber's salt), respectively. The figures show that Epsom salt releases half of its hydrated water at 150°C to 200°C and Glauber's salt releases half of its hydrated water at a much lower temperature. Figures 3-7 illustrate the thermogravimetric analysis of weight loss versus temperature for (hydrous aluminum silicate), and tale, hydrated clay zinc (zinc oxide/stearic acid, combinations acid oxide/carboxylic oxide/isononanoic acid, and zinc oxide/isooctanoic acid).

[0044] In the presence of water molecules, the OR groups of the grafted vinylsilane molecules can be easily hydrolyzed into OH groups. The Si-OH groups can then undergo a condensation reaction in the presence of a hydrolysis catalyst, for example dibutyltin dilaurate, to form Si-O-Si linkages. When there are not enough vinylsilane molecules grafted onto the elastomer backbone, the Si-O-Si linkages provide two dimensional chain extensions from the elastomer molecules. When there are enough vinylsilane molecules grafted onto the elastomer backbone, a three dimensional network can be formed, and the clastomers are crosslinked. The crosslinking process described above is shown below.